What is a mineral?

What are the five properties of a mineral?	
1.	
2.	
3.	
4.	
5.	
What is Arizona's state mineral?	
What is Arizona's state gem?	
Similarities and Differences between Rocks and	Minerals:
Similarities	Differences
•	•
•	•
•	•
Mineral I	Properties
r:	Luster:
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Caused by the trace elements that make up the mineral	 There are many types of luster, but there are two that are most commonly used

Why is color not a useful property?

Streak:

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Mineral Hardness:

- Scale created by the mineralogist Friedrich Mohs
- The scale was created to have one set method to determine the hardness of minerals studied and mined for because many people had different tests

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What are the tools that measure hardness and what hardness values do they represent?

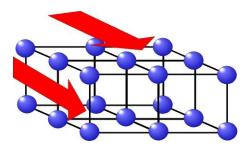
Cleavage:

In the youtube video clip, happened to the calcite when it was broken into several pieces? Were the pieces broken to similar shapes or completely irregular shapes?

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- Defined by planes from chemical make-up and crystal structure
- When broken using some force, the minerals displaying cleavage break along weak bonds
- Has different shapes

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When looking at a mineral with cleavage, you want to see whether it has planes or sides that are parallel on the opposite side of the prism or mineral.

Fracture:

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- Can break into curved, splintered, and jagged pieces
- Bonds are equal throughout mineral crystal structure

Quartz is the most common type of mineral that fractures.

Magnetism: Reaction with Acid

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- Mineral contain Iron (Fe), Nickel (Ni), or Colbot (Co)
- To test magnetism, use a magnet to see whether it is attracted to a mineral
- Example: Magnetite

• Example: Calcite

Density:

- Some minerals look the same
- Density can help distinguish minerals apart

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Example: Gold and Pyrite

Odor:

- Refers to smell
- Some minerals have a very distinctive smell
- Example: Sulfur

Taste:

- Some mineral have a specific taste
- Some examples are salty, sweet, etc
- Example: Halite

Double Refraction:

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- Produces a double image
- Often seen in clear minerals
- Example: Calcite

Fluorescence

- Glows in dark light
- Deals with wavelength of light trying to pass through the mineral
- Examples: Fluorite and Calcite

Radioactivity:

- Affected by chemical make up
- Some elements decay or lose particle to change the properties
- Use a Geiger counter to keep track of the decay rates