PERIODIC TABLE
ELEMENTS 1–20

Write the number of protons, electrons, and neutrons in each element.

HYDROGEN 1 # of Protons: # of Electrons: # of Neutrons: 1.01							**HELIUM** 2 # of Protons # of Electrons: # of Neutrons 4.00
LITHIUM 3 # of Protons # of Electrons: # of Neutrons 6.94	**BERYLLIUM** 4 # of Protons: # of Electrons. # of Neutrons 9.01	**BORON** 5 # of Protons: # of Electrons: # of Neutrons. 10.81	**CARBON** 6 # of Protons: # of Electrons: # of Neutrons 12.01	**NITROGEN** 7 # of Protons # of Electrons # of Neutrons 14.01	**OXYGEN** 8 # of Protons # of Electrons # of Neutrons: 16.00	**FLUORINE** 9 # of Protons: # of Electrons # of Neutrons 19.00	**NEON** 10 # of Protons # of Electrons # of Neutrons: 20.18
SODIUM 11 # of Protons: # of Electrons. # of Neutrons: 22.99	**MAGNESIUM** 12 # of Protons: # of Electrons # of Neutrons: 24.31	**ALUMINUM** 13 # of Protons: # of Electrons. # of Neutrons: 26.98	**SILICON** 14 # of Protons: # of Electrons # of Neutrons: 28.09	**PHOSPHORUS** 15 # of Protons. # of Electrons. # of Neutrons: 30.97	**SULFUR** 16 # of Protons: # of Electrons # of Neutrons: 32.07	**CHLORINE** 17 # of Protons. # of Electrons # of Neutrons: 35.45	**ARGON** 18 # of Protons: # of Electrons # of Neutrons: 39.95
POTASSIUM 19 # of Protons: # of Electrons # of Neutrons. 39.10	**CALCIUM** 20 # of Protons: # of Electrons. # of Neutrons: 40.08						

Note: Remember that the number of neutrons is not the same for every atom of an element. The number of neutrons you write in this chart will be a number, that when added to the number of protons, gives a sum as close as possible to the atomic mass.

Name _____

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