

# Periodic Table – Special Families

Name: \_\_\_\_\_

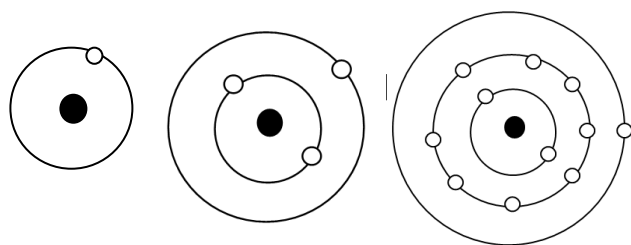
Date: \_\_\_\_\_

Period: \_\_\_\_\_ Page: \_\_\_\_\_

Groups/Columns/Families \_\_\_\_\_, \_\_\_\_\_

Group 1A: \_\_\_\_\_

\_\_\_\_\_ electrons in the \_\_\_\_\_ level



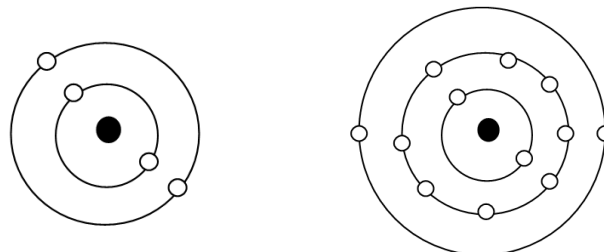
Hydrogen – 1

Lithium – 3

Sodium – 11

Group 2A: \_\_\_\_\_

\_\_\_\_\_ electrons in the \_\_\_\_\_ level

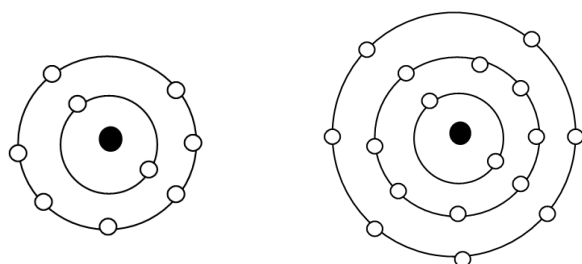


Beryllium – 2

Magnesium – 12

Group 7A: \_\_\_\_\_

\_\_\_\_\_ electrons in the \_\_\_\_\_ level

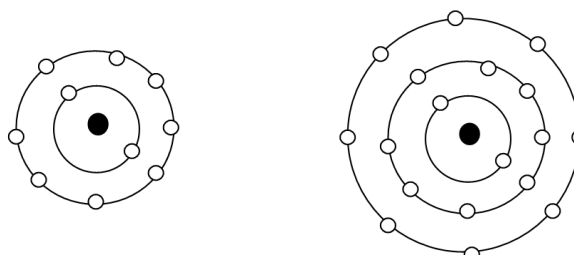


Fluorine – 9

Chlorine – 17

Group 8A: \_\_\_\_\_

All have \_\_\_\_\_ levels



Neon – 10

Argon – 18

Stable Atoms: \_\_\_\_\_

All Atoms try to become “stable” = \_\_\_\_\_

Two kinds of bonds: Gain or lose electrons = \_\_\_\_\_ bond

Share electron = \_\_\_\_\_ bond