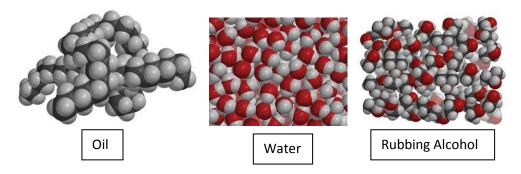
## **Sink or Float: Density of Liquids Key Concepts:** Since density is a characteristic property of a substance, each liquid has its own The \_\_\_\_\_ of a liquid determines whether it will \_\_\_\_\_ or \_\_\_\_\_ in another liquid. A liquid will float if it is \_\_\_\_\_\_ than the liquid it is placed in. A liquid will sink if it is \_\_\_\_\_\_ than the liquid it is placed in. **Sink or Float: Density of Liquids Key Concepts:** Since density is a characteristic property of a substance, each liquid has its own The \_\_\_\_\_ of a liquid determines whether it will \_\_\_\_\_ or in another liquid. A liquid will float if it is \_\_\_\_\_\_ than the liquid it is placed in. A liquid will sink if it is \_\_\_\_\_\_ than the liquid it is placed in. **Sink or Float: Density of Liquids Key Concepts:** Since density is a characteristic property of a substance, each liquid has its own The \_\_\_\_\_ of a liquid determines whether it will \_\_\_\_\_ or in another liquid. A liquid will float if it is \_\_\_\_\_\_ than the liquid it is placed in.

A liquid will sink if it is than the liquid it is placed in.

## **Sink or Float: Density of Liquids Processing:**

The pictures below are of the liquids used in the lab at a molecular level.



Describe, on a molecular level, why the oil was less dense compared to the water?

Describe, on a molecular level, why the water was the densest liquid out the three used in the lab?

Describe, on a molecular level, why rubbing alcohol was the least dense liquid used in the lab?