## Procedure:

- 1. Use paper clip to make a looped end wire.
- 2. Dip loop in chemical and hold in flame.
- 3. Write down the color of the flame.
- 4. Repeat while looking through a diffraction grating spectrometer.
- 5. Record the areas where you see a dark band or bands of color.
- 6. Dip loop in water and wipe clean.
- 7. Repeat the steps for all the chemicals.
- 8. Clean up and return all materials and equipment to proper place.

## Data:

Substance (chemical formula)	Observations of Chemical	Color of Flame
Calcium Chloride (CaCl)		
Lithium Chloride (LiCl)		
Strontium Chloride (SrCl)		
Potassium Chloride (KCl)		
Sodium Chloride (NaCl)		
Cupric Chloride (CuCl <sub>2</sub> )		
Unknown # 1		

Colors using the Spectroscope						
Source	Red	Orange	Yellow	Green	Blue	Violet
Calcium Chloride (CaCl)						
Lithium Chloride (LiCl)						
Strontium Chloride (SrCl)						
Potassium Chloride (KCl)						
Sodium Chloride (NaCl)						
Cupric Chloride (CuCl <sub>2</sub> )						

1) How do scientists know what different stars are composed of without visiting them?